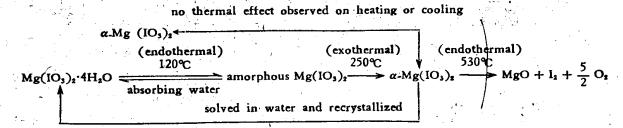
THE STRUCTURE AND PHASE TRANSITIONS OF MAGNESIUM IODATE CRYSTALS

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ABSTRACT

The phase transitions of Mg(IO_s). 4H₂O crystals in the heating process have been investigated by means of thermal analyses and X-ray diffraction at high and room temperatures. The results may be exhibited in the following diagram:



The crystal structure of Mg(IO₃), 4H₂O has been determined by means of the method of single crystal diffraction, it has been found that the space group of Mg(IO₃), 4H₂O is P2 or Pm. There are two formula units per unit cell. Lattice parameters are: a = 8.307Å, b = 6.627Å, c = 8.541Å, $\beta = 100^{\circ}45'$.

The crystal structure of a-Mg(IO₃), h s been determined by X-ray powder method. The unit cell of a-Mg(IO₃), is hexagonal, with a =5.4777Å, c = 5.1282Å, c/a = 0.9362 at 14°C. There is one formula unit per unit cell. The space group is P6₃. 60° are situated at the 6(c) positions, 21° are situated at 2(b) positions, and Mg⁺⁺-ion is disorderly distributed at half of 2(a) positions, the atomic parameters being $X_c = 0.096$, $Y_c = 0.344$, $Z_c = 0.162$, $Z_b = 0$, $Z_a = 0.981$. The crystal structure of a-Mg(IO₃), is isomorphic with a-LiIO₃. a-Mg(IO₃), is different from a-LiIO₃ only in that the Mg⁺⁺-ion is disorderly distributed at one-half of 2(a) positions and the other half is vacant.